

AFM Application Problems Unit 5

Round to the hundredths place when necessary and do not forget units.

1. A bacterial culture doubles every 2 hours. If the culture started with 24,000 bacteria, how many bacteria will be present in 5 hours?

$$A = 24,000 (2)^{5/2}$$

$$A = 135,765 \text{ bacteria rounded up}$$

2. The half life of a radioactive sample is 4 hours. If 60 g of the sample was initially present, how much will remain after 7 hours?

$$A = 60 (1/2)^{7/4}$$

$$A = 17.84 \text{ g}$$

3. The population of a town triples every 6 years. If 4000 people are present in 2006, how many people will be in the town in 2016?

$$A = 4000 (3)^{10/6}$$

$$A = 24,962 \text{ people}$$

4. The half life of a radioactive sample is 6.2 hours. If 2000 g of the sample is present after 7 hours, how much was initially present?

$$2000 = P (1/2)^{7/6.2}$$

$$P = 4,375 \text{ g}$$

5. A bacterial culture doubles every 5 hours. How long will it take for a culture to quadruple?

$$4 = (2)^{t/5} \quad t = 10 \text{ hours}$$

6. A radioactive sample has a half life of 3 days. How long will it take for only 1/8 of the sample to remain?

$$1/8 = (1/2)^{t/3} \quad t = 9 \text{ days}$$

7. The population of a town triples every 8 years. How many years will it take for the population to double?

$$5.05 \text{ years}$$

$$\frac{\ln 3}{8} = k$$

$$2 = e^{\frac{\ln 3}{8} \cdot t}$$

8. The population of a town halves every 15 years. In how many years will 20% of the population have fled?

$$.2 = (1/2)^{t/15}$$

$$t = \frac{15 \log .2}{\log .5} \quad t \approx 4.83 \text{ years}$$

9. A radioactive sample has a half life of 3 years, and has an initial mass of 68 g. How many months will it take for the sample to lose 8 g?

$$60 = 68 (1/2)^{t/36}$$

$$\log \frac{60}{68} = \frac{t}{36} \log (1/2) \quad t \approx 6.5 \text{ months}$$

10. \$5000 is invested at 7.2% compounded annually for 4 years. What is the amount of money at the end of the 4 years?

$$A = 5000 (1 + .072)^4$$

$$A = \$6,603.12$$

11. \$2300 is invested at 6% compounded monthly for 7 years. How much interest is earned?

$$A = 2300 (1 + \frac{.06}{12})^{84}$$

$$A = \$3496.85$$

$$\text{Interest} = \$1196.85$$

12. \$300 is invested at 10% compounded quarterly. How many years must it stay in the bank to double?

$$2 = \left(1 + \frac{.10}{4}\right)^{4t} \quad t = \frac{\log 2}{4 \log 1.025}$$
$$\log 2 = 4t \log(1.025) \quad t = 7.02 \text{ years}$$

13. An investment is invested at 5% compounded continuously. How many years will it take to triple in value?

$$3 = e^{.05t}$$
$$\ln 3 = .05t \quad t = \frac{\ln 3}{.05} \quad t = 21.97 \text{ years}$$

14. A small tremor of magnitude 3.4 is followed by a stronger one of magnitude 4.1. How much stronger is the second tremor than the first?

$$= 10^{4.1-3.4}$$
$$= 10^{.7} \quad \approx 5.01$$

15. An earthquake has a magnitude of 6.5, and the following day a stronger earthquake occurs with double the intensity. What is the magnitude of the stronger earthquake?

$$2 = 10^{x-6.5}$$
$$\log 2 + 6.5 = x \quad x \approx 6.8$$

16. A strong earthquake with a magnitude of 6.7 is three times more intense than a weaker earthquake. What is the magnitude of the weaker earthquake?

$$3 = 10^{6.7-x}$$
$$\log 3 = (6.7-x) \log 10$$
$$x = 6.7 - \log 3 \quad x \approx 6.22$$

17. A small tremor of magnitude 2.3 is then followed by a stronger one of magnitude 5.3. How much stronger is the second tremor than the first?

$$10^{5.3-2.3}$$
$$= 10^3 \quad 1,000 \text{ times}$$

18. An earthquake has a magnitude of 6.7, and the following week a stronger earthquake occurs with six times the intensity. What is the magnitude of the stronger earthquake?

$$6 = 10^{x-6.7}$$
$$\log 6 + 6.7 = x \quad x \approx 7.48$$

19. An earthquake with a magnitude of 5.9 is one thousand times more intense than a weaker earthquake. What is the magnitude of the weaker earthquake?

$$1000 = 10^{5.9-x}$$
$$x = 2.9$$

20. The formula for acid strength is: $\text{pH} = -\log[\text{H}^+]$, where PH is the acid strength and H^+ is the concentration of hydrogen ion. A beaker of acid has a hydrogen concentration of 3.5×10^{-6} mol/L. Calculate the pH of the acid.

$$\text{pH} = -\log[3.5 \times 10^{-6}]$$
$$\text{pH} = 5.46$$